

**GHS Product Identifier:** 101.07, Potassium Iodide Technical; 101.24, Potassium Iodide USP; 101.14, Potassium Iodide ACS; 101.08, Potassium Iodide 45% Soln.; 101.09, Potassium Iodide 50% Soln.; 101.84, Potassium Iodide Photo Grade.

**Formula Description: Technical:** Off white to light brown crystals or granular powder. **USP/ACS/Photo:** Colorless or white crystals or granular powder; slightly hygroscopic in moist air. Tends to cake during storage. On long exposure to air becomes yellow due to liberation of iodine. Light and moisture accelerates decomposition. **45% & 50% Solution:** Clear colorless to light yellow solution.

**Recommended Use:** Potassium Iodide is an inorganic halogenated salt that is used in polymer industry in improving structural properties<sup>1-2</sup>. It is a corrosion inhibitor/acid intensifier in oilfield gas production<sup>3-4</sup>, used in x-ray films owing to luminescence properties<sup>5</sup>, LCD manufacturing as a polarizer<sup>6</sup>, nylon stabilizer<sup>7</sup>, trace mineral in animal feeds and/or dietary supplement and food additive<sup>8</sup>.

**General Properties: Technical/USP/ ACS**

<b>Molecular Weight</b>	<b>166.0</b>	<b>Density</b>	<b>3.12 (25°C)</b>
<b>Solubility</b>	<b>144 g/100 ml H<sub>2</sub>O (20°C)</b>	<b>Solubility</b>	<b>208 g/100 ml H<sub>2</sub>O (100°C)</b>

**General Properties: 45% & 50% Solution**

<b>45% Solution</b>	<b>Density</b>	<b>12.2 lbs/gal</b>	<b>1.46 g/ml</b>
<b>50% Solution</b>	<b>Density</b>	<b>12.77 lbs/gal</b>	<b>1.54 g/ml</b>

**General Properties: Photo Grade**

<b>Molecular Weight</b>	<b>166.0</b>	<b>Composition</b>	<b>Iodine 76.45%</b>	<b>Potassium 23.55%</b>
<b>Solubility</b>	<b>144 g/100 ml H<sub>2</sub>O (20°C)</b>	<b>Solubility</b>	<b>208 g/100 ml H<sub>2</sub>O (100°C)</b>	

**Chemical Product Specifications**

	<b>Tech</b>
<b>Assay</b>	98.0% min

	<b>45% Solution</b>	<b>50% Solution</b>
<b>Assay</b>	44.5% - 45.5%	49.5% - 50.4%
<b>pH (as is)</b>	7.0 - 11.0	7.0 - 11.0

Deepwater's **PurIzty** products offer you full traceability for all raw materials.

All products are manufactured under current Good Manufacturing Practices (cGMP)

in our US FDA registered plant. FEI #2013633.



	<b>PurIzty USP</b>	<b>PurIzty ACS</b>	<b>PurIzty Photo</b>
<b>Assay</b>	99.0% – 101.5% (Anhydrous)	99.0% min (as is)	99.0% min (as is)
<b>Identification</b>	USP Standards A&B		
<b>Alkalinity</b>	USP Standards		
<b>pH (5% Solution)</b>		6.0 – 9.2	6.0 – 9.2
<b>Insoluble Matter</b>		0.005% max	0.005% max

<b>Loss on Drying</b>	1.0% max	0.2% max	0.2% max
	<b>PurIzty USP</b>	<b>PurIzty ACS</b>	<b>PurIzty Photo</b>
<b>Chloride &amp; Bromide (as Cl)</b>		0.01% max	0.01% max
<b>Iodate (IO<sub>3</sub>)</b>	4 ppm max	3 ppm max	3 ppm max
<b>Nitrogen Compounds (as N)</b>			0.001% max
<b>Phosphate (PO<sub>4</sub>)</b>		0.001% max	0.001% max
<b>Sulfate (SO<sub>4</sub>)</b>		0.005% max	0.005% max
<b>Nitrate, Nitrite &amp; Ammonia</b>	USP Standards		
<b>Thiosulfate &amp; Barium</b>	USP Standards		
<b>Barium (Ba)</b>		0.002% max	0.002% max
<b>Heavy Metals (as Pb)</b>		5 ppm max	5 ppm max
<b>Iron (Fe)</b>		3 ppm max	3 ppm max
<b>Calcium (Ca)</b>		0.002% max	0.002% max
<b>Magnesium (Mg)</b>		0.001% max	0.001% max
<b>Sodium (Na)</b>		0.005% max	0.005% max
<b>Elemental Impurities Class 1</b>	Cd, Pb, As, Hg		
<b>Elemental Impurities Class 2A</b>	Co, V, Ni		

\*Compendial grades conform to current USP and ACS editions

### Standard Packaging

Net Weight	Packaging	Product
50 lbs.	LDPE 3 gal Pail	Tech Only
25 lbs.	LDPE 2 gal Pail	USP/ACS
100 lbs.	UN1G 8 gal Fiberdrum	USP/ACS
65 lbs.	5 gal HDPE Drum	45% Soln
600 lbs.	55 gal HDPE Drum	45% Soln
650 lbs.	55 gal HDPE Drum	50% Soln
Material packaged with Saran inner liner and polyethylene out liner; suitable for export. Curtec drum does not include liner.		

SDS with detailed information available upon request.

### References:

1. Nadimicherla R, Kalla R, Muchakayala R, Guo X (2015) Effects of potassium iodide (KI) on crystallinity, thermal stability, and electrical properties of polymer blend electrolytes (PVC/PEO: KI). Solid State Ionics 278:260–267
2. Solid State Communications Volume 149, Issues 31–32, August 2009, Pages 1282-1287 Effect of nanofillers on thermal and transport properties of potassium iodide–polyethylene oxide solid polymer electrolyte
3. Journal of Applied Electrochemistry August 2004, Volume 34, Issue 8, pp 833–839 Synergistic Influence of Poly(4-Vinylpyridine) and Potassium Iodide on Inhibition of Corrosion of Mild Steel in 1M HCl L. Larabi Y. Harek M. Traisnel A. Mansri
4. Salah Merah, Lahcene Larabi, Omar Benali, Yahia Harek, (2008) "Synergistic effect of methyl red dye and potassium iodide on inhibition of corrosion of carbon steel in 0.5 M H<sub>2</sub>SO<sub>4</sub>", Pigment & Resin Technology, Vol. 37 Issue: 5, pp.291-298
5. An X-Ray Study of Formamide and Solutions of Potassium Iodide in Formamide R. DeSando, and G. Brown J. Phys. Chem., 1968, 72 (4), pp 1088–1091

6. A Study on the Heat Resistance and Polarization Characteristics of Poly(vinyl alcohol)-I<sub>2</sub> Complex Films Prepared with a Potassium iodide Applied Chemistry for Engineering  
Volume 10 Issue 4 / Pages.603-607 / 1999 / 1225-0112(pISSN) / 2288-4505(eISSN) Oh, Se Young ;  
Shin, Dong Yoon
7. Structure of iodide ions in iodinated nylon 6 and the evolution of hydrogen bonds between parallel chains in nylon 6N. S. Murthy, Macromolecules, 1987, 20 (2), pp 309–316
8. Waszkowiak, K. and Szymandera-Buszk, K. (2008), Effect of storage conditions on potassium iodide stability in iodised table salt and collagen preparations. International Journal of Food Science & Technology, 43: 895-899.

